

Rydberg Atoms

INTRODUÇÃO À FÍSICA ATÔMICA E MOLECULAR
INSTITUTO DE FÍSICA DE SÃO CARLOS-USP

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Contents

1. Introduction

2. Exciting Rydberg states

3. Rydberg states wave function

4. Quantum processor

5. Conclusion

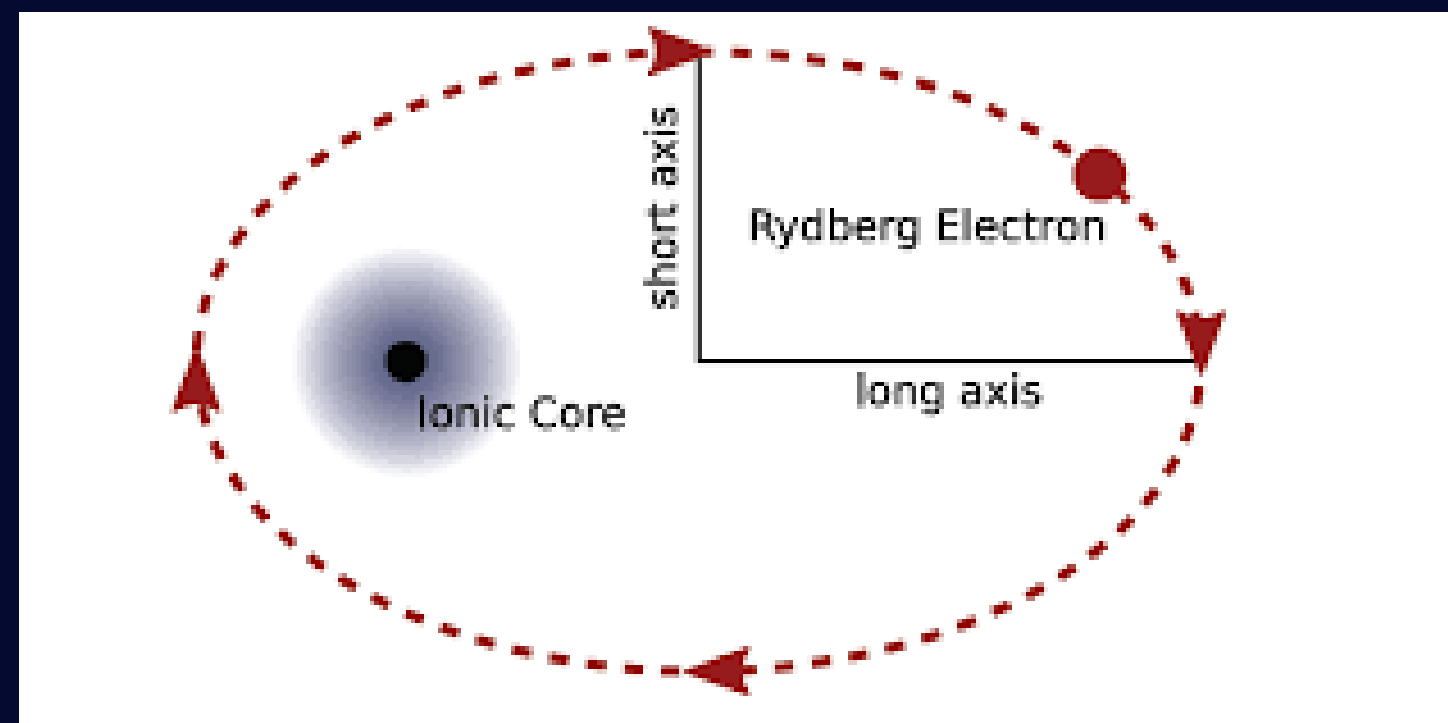
6. References

1. Introduction

1. Introduction

1.1. What are Rydberg atoms?

- Atoms with an electron in a state with high principal quantum number n .
 - Good coupling with fields.
 - Long range dipole-dipole interactions.



1. Introduction

1.2. History

- 1890 - Empirical equation to several frequencies in the hydrogen spectrum (Johannes Rydberg).

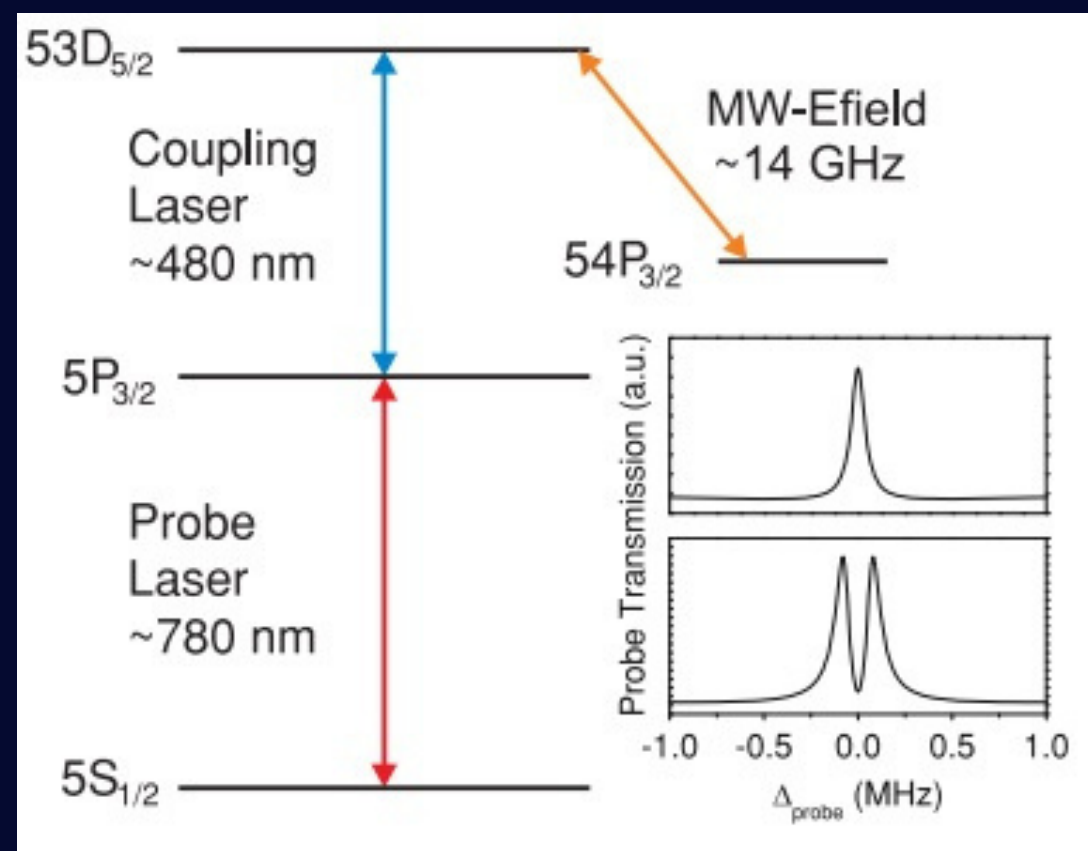
$$f = R_H \left(\frac{1}{m^2} - \frac{1}{n^2} \right)$$

- Bohr and Som-merfeld, quantizing the angular momentum of the electrons' orbits.

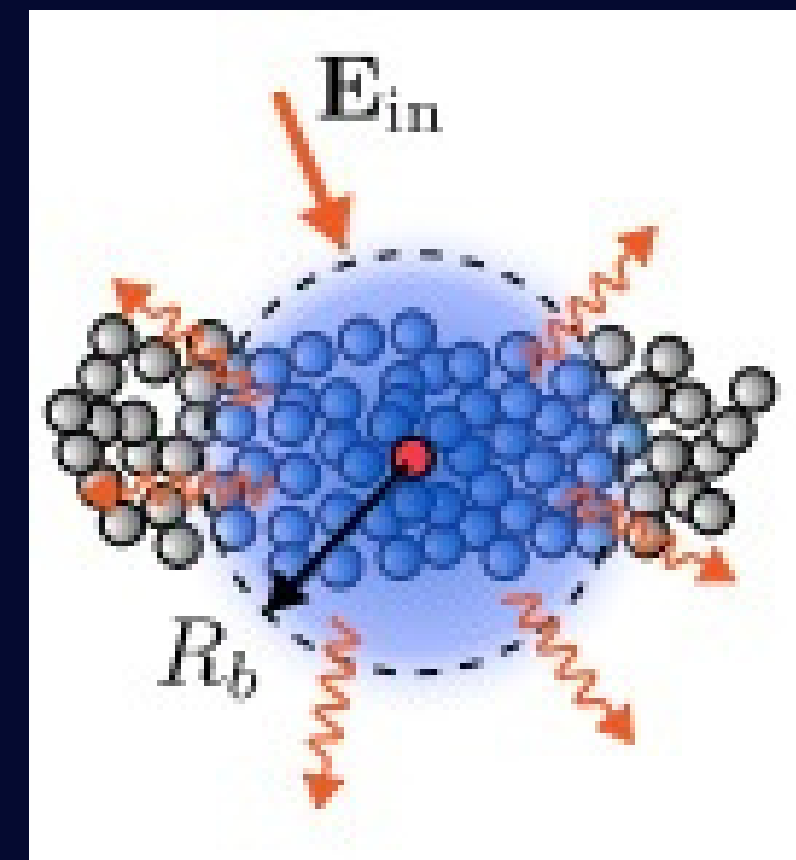
1. Introduction

1.3. Applications

High sensibility sensors*



Nonlinear optics**



*H. Fan et al., J. Phys. B At. Mol. Opt. Phys. 48, 202001 (2015)

**M. Moreno-Cardoner, D. Goncalves, e D. E. Chang, Phys. Rev. Lett. 127, 263602 (2021).

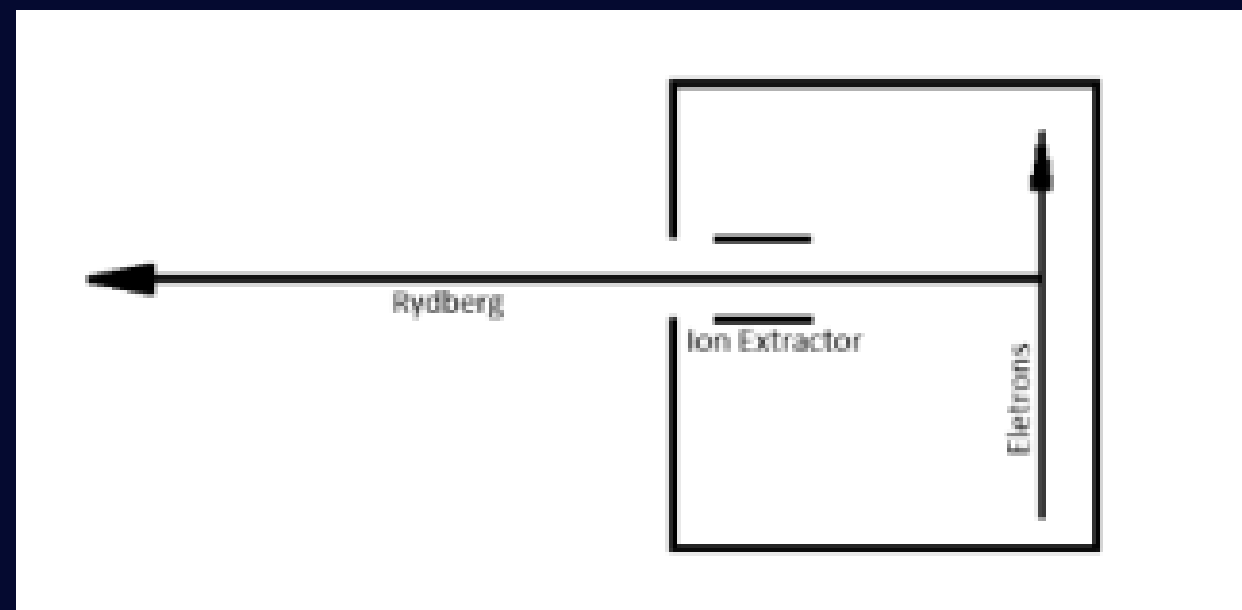
2. Exciting Rydberg states

2. Exciting Rydberg states

2.1. Electron impact

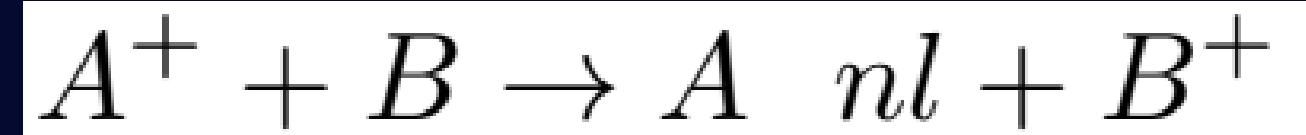


Transferring energy from the electrons beam to the valence electron on the atom.

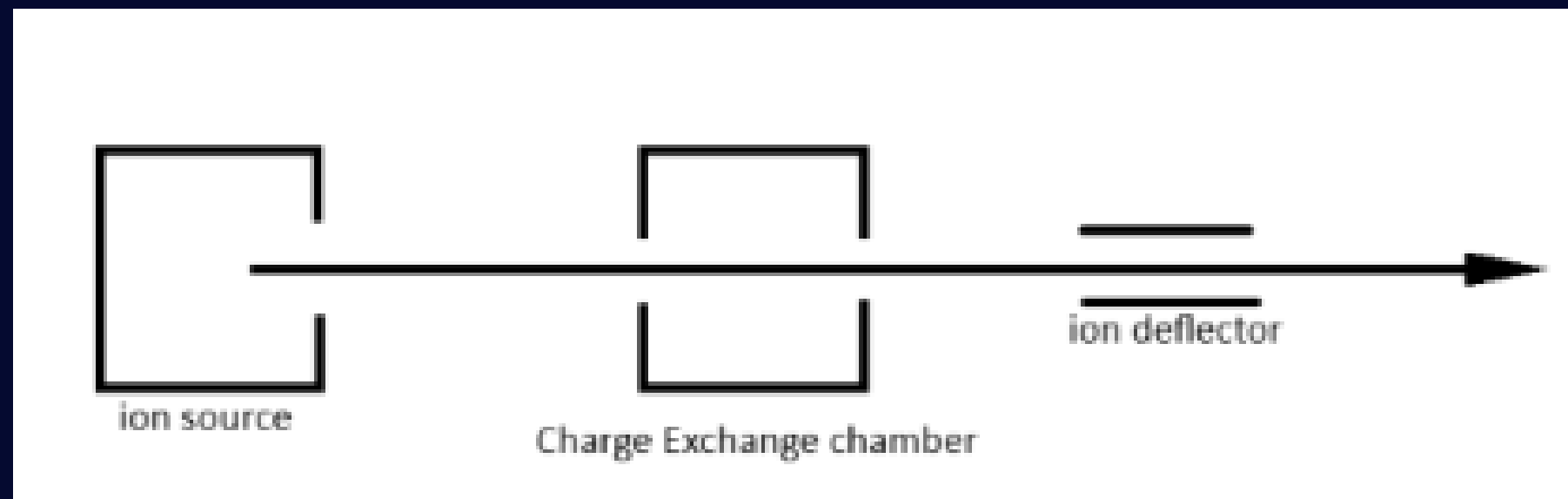


2. Exciting Rydberg states

2.2 Charge exchange



Ions in the ion beam capture electrons from the rare gases in the charge exchange chamber.

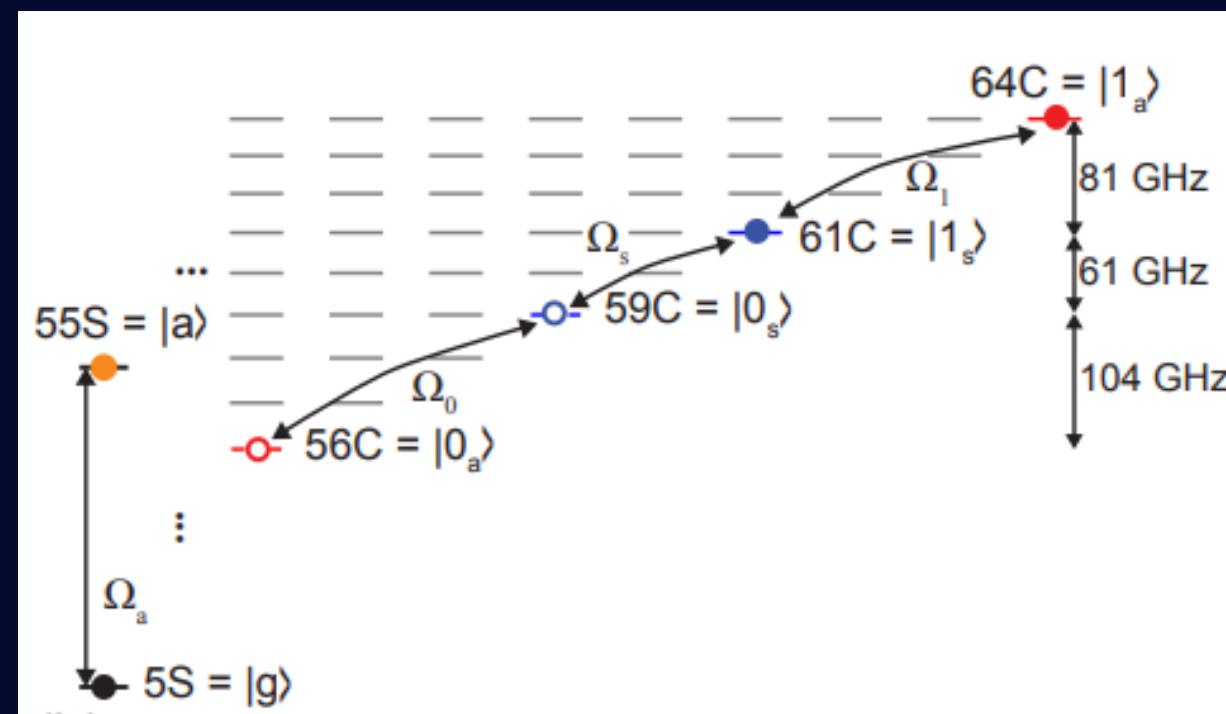


2. Exciting Rydberg states

2.3. Optical excitation

$$hf + A \rightarrow A \quad nl$$

A photon in resonance with the atomic transition is absorbed, and this energy causes the atom to transition to a higher energy state.



3. Rydberg states wave function

3. Rydberg states wave function

Hydrogen like Hamiltonian:

$$\left(-\frac{\nabla^2}{2} - \frac{1}{r} \right) \psi = W \psi$$

Should work with atoms with a single valence electron far from the nucleus.

Angular solution:

$$Y_{lm}(\theta, \phi) = \sqrt{\frac{(l-m)!}{(l+m)!} \frac{2l+1}{4\pi}} P_l^m(\cos \theta) e^{im\phi}$$

*T. F. Gallagher, Cambridge monographs on atomic, molecular and chemical physics: Rydberg atoms series number 3 (Cambridge University Press, Cambridge, England, 2011).

3. Rydberg states wave function

The radial equation can be written as:

$$\frac{\partial^2 \rho}{\partial r^2} + \left[2W + \frac{2}{r} - \frac{l(l+1)}{r^2} \right] \rho = 0$$

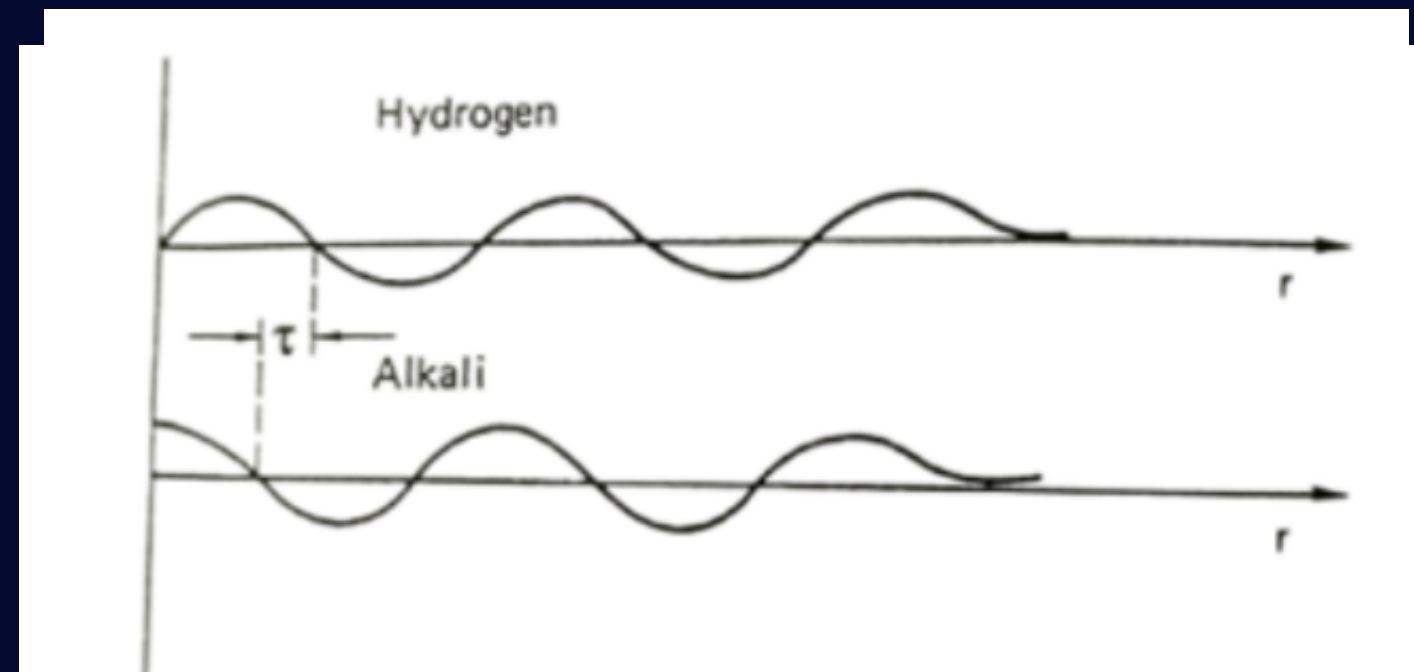
Which is the Coulomb equation. For our problem, we can consider two solutions, but taking into account the boundary conditions, only the regular solution remains. This can be written as:

$$f \rightarrow u(\beta, l, r) \sin \pi\beta - v(\beta, l, r) e^{i\pi\beta}$$

*M. J. Seaton, Rep. Prog. Phys. 46, 167 (1983).

3. Rydberg states wave function

We now, need to make a correction for an electron that has a considerable probability of being found near the ionic nucleus, obtaining a phase due to the action of the new potential.



it is possible to express this phase as:

$$\tau = \int_0^{r_0} \left\{ [W - V_{eff}(r)]^{\frac{1}{2}} - \left[W + \frac{1}{r} \right]^{\frac{1}{2}} \right\} \sqrt{2} \, dr$$

3. Rydberg states wave function

We can simplify this equation by considering only the difference between the potentials :

$$\tau = \int_0^{r_0} V_d(r) \left(\frac{r}{2}\right)^{\frac{1}{2}} dr.$$

Thus, the new solution to the radial equation in terms of the phase will be given by:

$$\rho(r) = f(W, l, r) \cos \tau - g(W, l, r) \sin \tau$$

3. Rydberg states wave function

Considering the new boundary conditions, we can define the new wave function, with the potential correction for the electron near the ionic nucleus.

$$\psi = \frac{Y_{lm}[f(W, l, r) \cos \pi \delta_l - g(W, l, r) \sin \pi \delta_l]}{r}$$

In which the allowed energies are given by:

$$W = -\frac{1}{2(n - \delta_l)^2}$$

And the quantum defects for each state with quantum number l are given by:

$$\delta_l = \frac{\tau}{\pi}$$

4. Quantum processor

4. Quantum processor

A quantum processor based on coherent transport of entangled atom arrays

Dolev Bluvstein¹, Harry Levine^{1,†}, Giulia Semeghini¹, Tout T. Wang¹,
Sepehr Ebadi¹, Marcin Kalinowski¹, Alexander Keesling^{1,2}, Nishad Maskara¹,
Hannes Pichler^{3,4}, Markus Greiner¹, Vladan Vuletić⁵, and Mikhail D. Lukin¹

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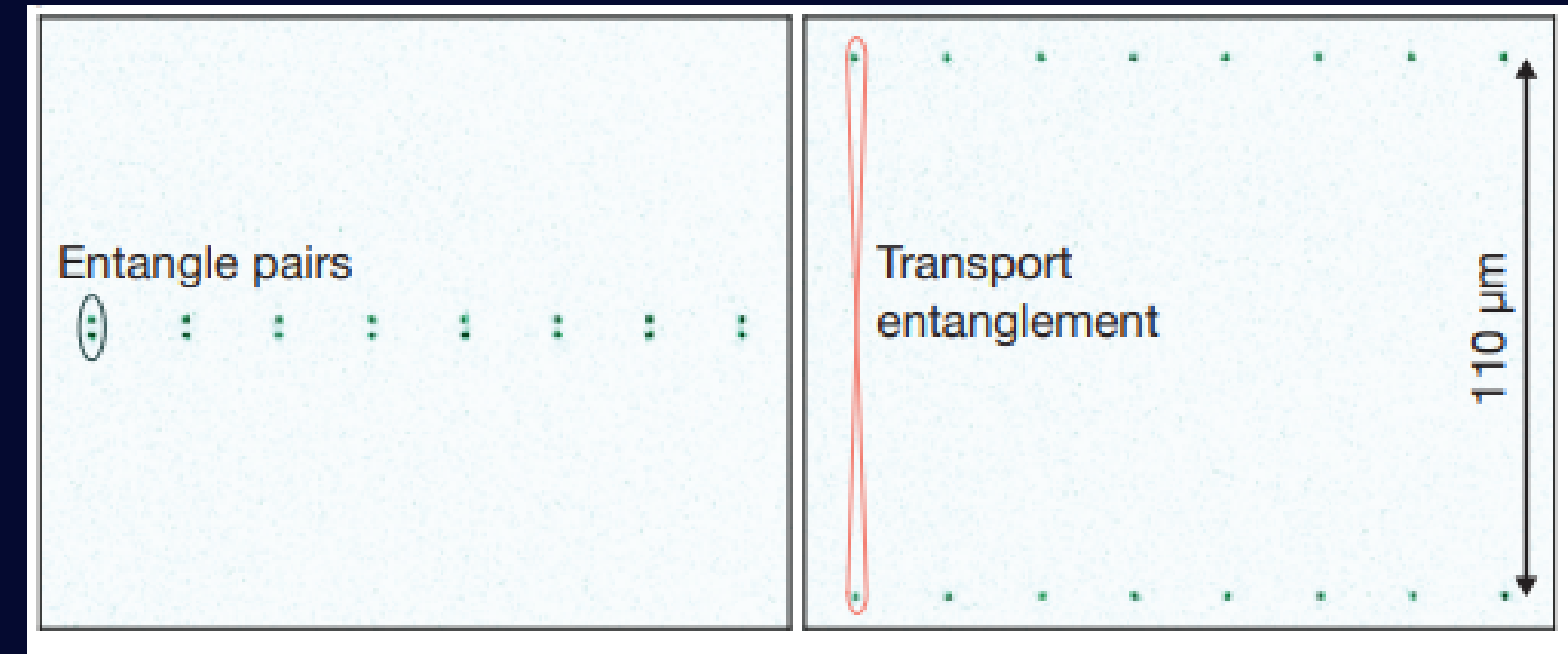
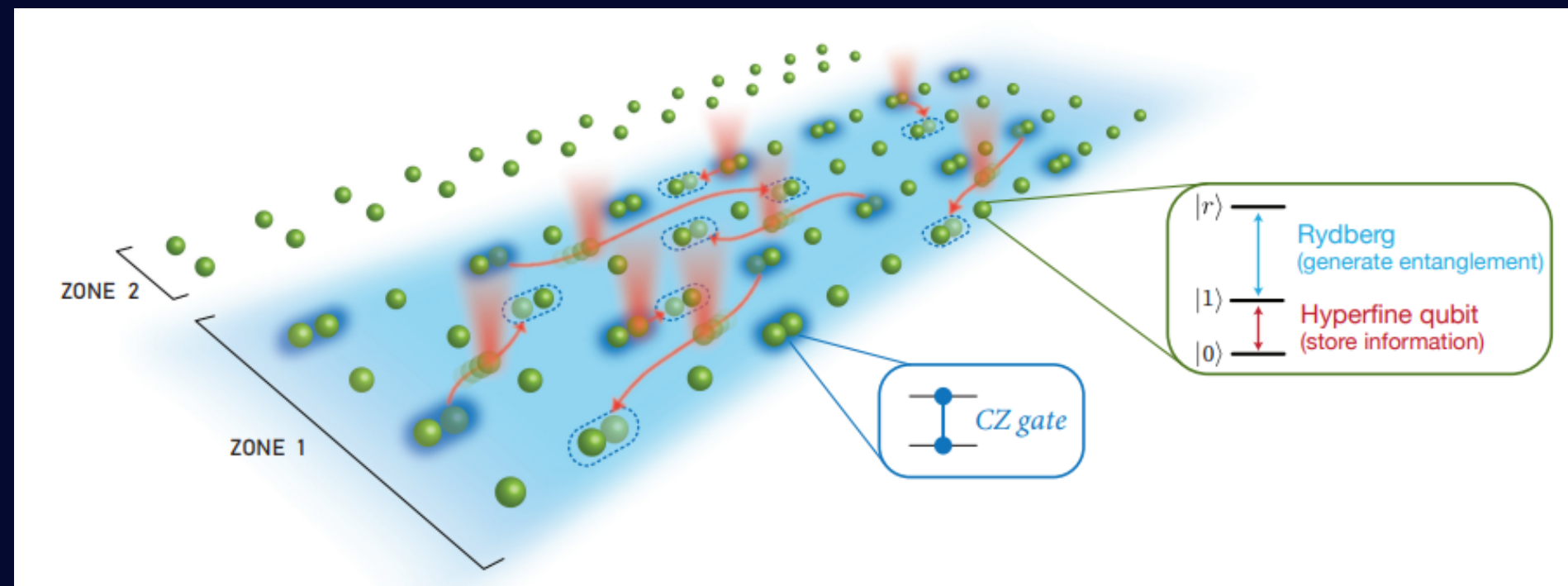
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[†]*Current affiliation: AWS Center for Quantum Computing, Pasadena, CA 91125.*

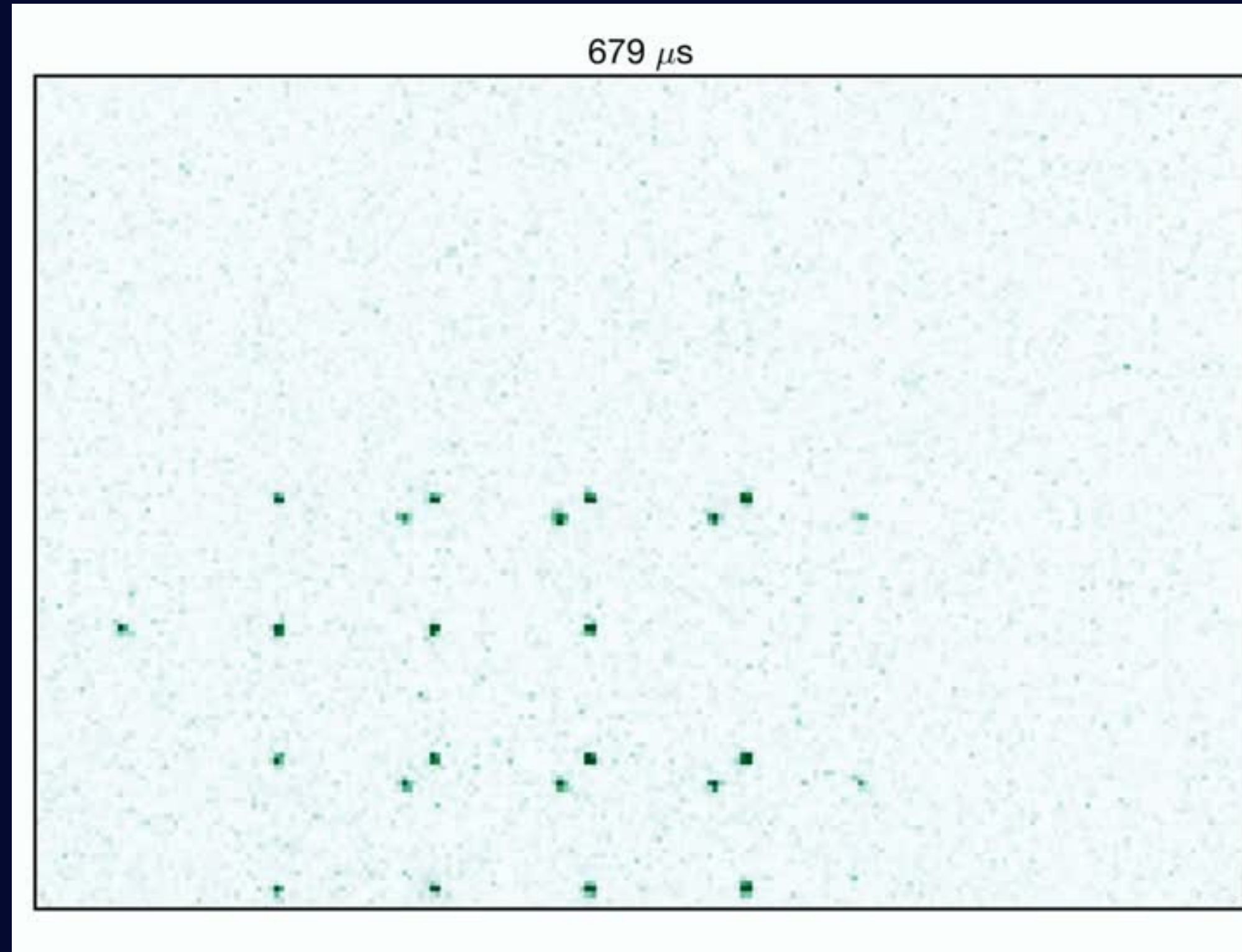
This is a cold atom experiment, located at Harvard University, they have a system of optical tweezers capable of transporting atoms with high precision.

4. Quantum processor

They use the long-range interactions of Rydberg states to generate entanglement between atoms, making CZ gates in parallel in hyperfine states $\{|0\rangle$ and $|1\rangle\}$ using rydberg transition $|1\rangle \rightarrow |r\rangle$.



4. Quantum processor



5. Conclusion

5. Conclusion

- Similarity between the wave function of Rydberg atoms with a single valence electron and the wave function of hydrogen. This gives a good understanding of the wave function of Rydberg states
- Rydberg atoms are highly applicable in quantum technologies, and their “exaggerated” characteristics can be exploited in a variety of ways.

6. References

- [1] Jonathon Sedlacek Harald Kübler Shaya Karimkashi Haoquan Fan, Santosh Kumar and James P Shaffer. Atom based rf electric field sensing. J. Phys. B: At. Mol. Opt. Phys. 48, 2015.
- [2] D. Goncalves M. Moreno-Cardoner and D. E. Chang. Quantum nonlinear optics based on two-dimensional rydberg atom arrays. Phys. Rev. Lett. 127, 263602, 2021
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